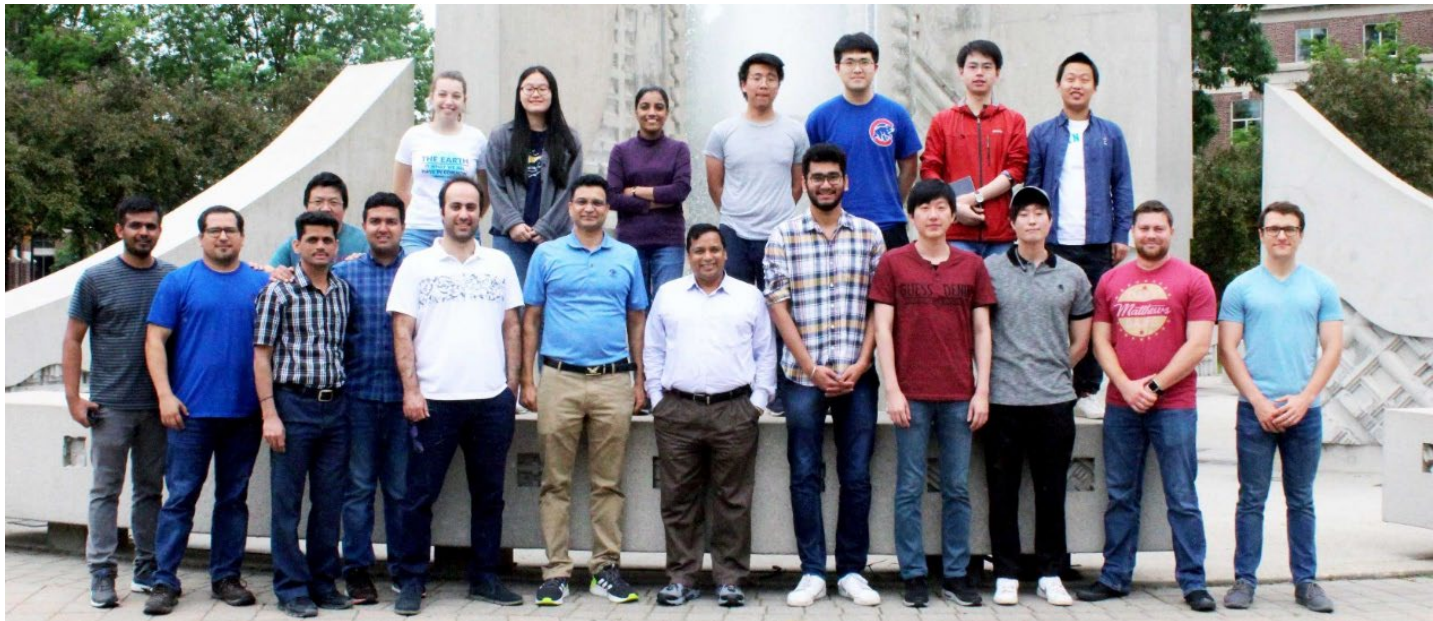


# Li-ion Batteries: Thermally Safe and Early Sensing

**Vilas G. Pol**





# THE NOBEL PRIZE IN CHEMISTRY 2019

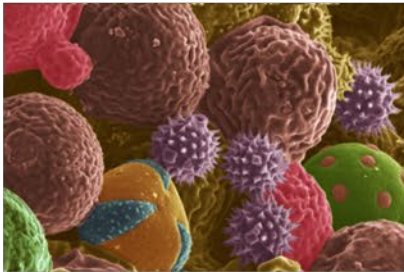


**John B. Goodenough, M. Stanley Whittingham, Akira Yoshino**

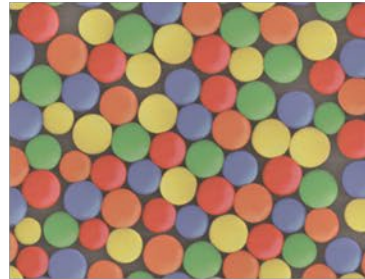
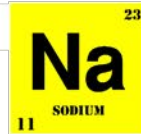
# Purdue's Research on Battery Technologies



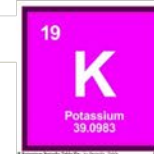
## Lithium-ion



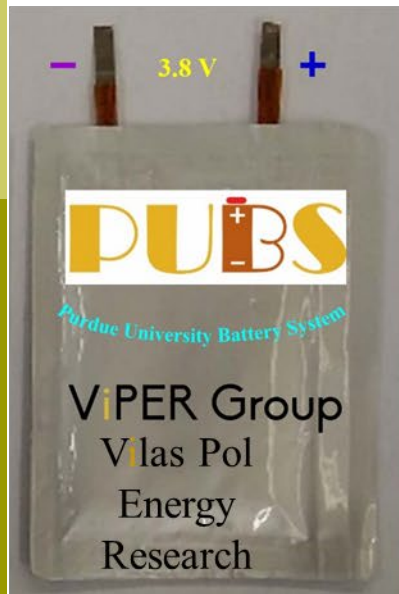
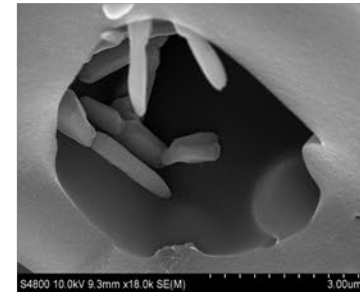
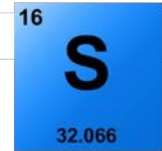
## Sodium ion



## Potassium ion



## Li-S/Solid state



*Interest: Batteries for Renewable Energy Storage and their Safety*



PURDUE  
UNIVERSITY



SBIR/STTR  
SMALL BUSINESS INNOVATION RESEARCH  
SMALL BUSINESS TECHNOLOGY TRANSFER



TIMKEN



PURDUE  
RESEARCH FOUNDATION

PURDUE  
UNIVERSITY

Purdue Process Safety and Assurance Center

**225 Peer-reviewed publications and 30 US Patents/applications**

# Battery Research Challenges



- **Cost**

Current projected cost (25 kW battery) ~ \$1000

- Target cost (25 kW battery) ~ \$500



- **Safety**

Inherently safe batteries needed

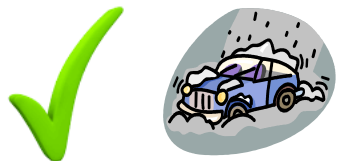
- Overcharge protection circuitry expensive



- **Life**

Current technology ~ 5 to 10 years

- Target ~ 15 years



- **Low Temperature Performance**

Current technology ~ Sluggish  $< 0^{\circ}\text{C}$

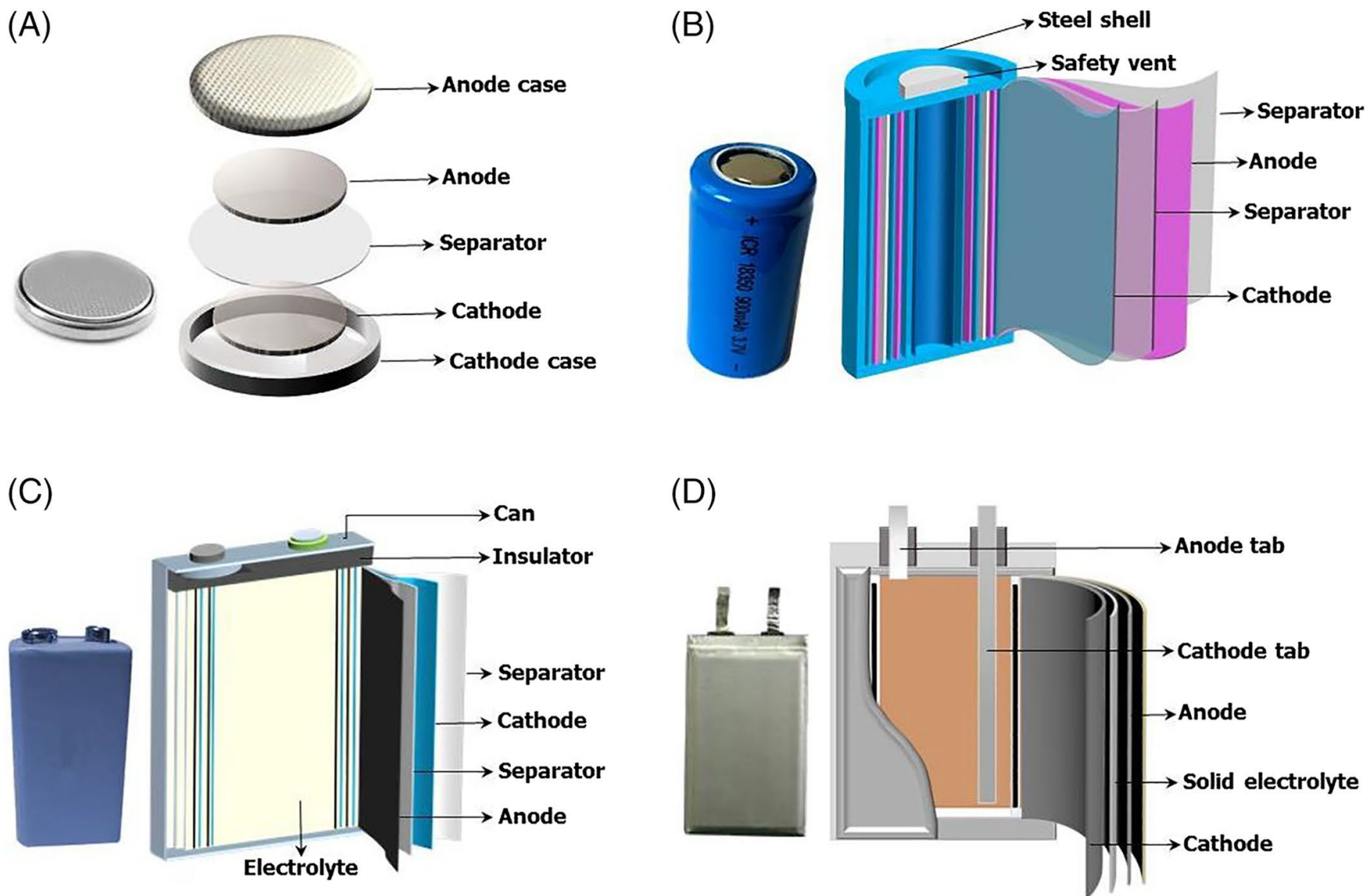
- Target ~  $-30^{\circ}\text{C}$  (cold cranking)

New

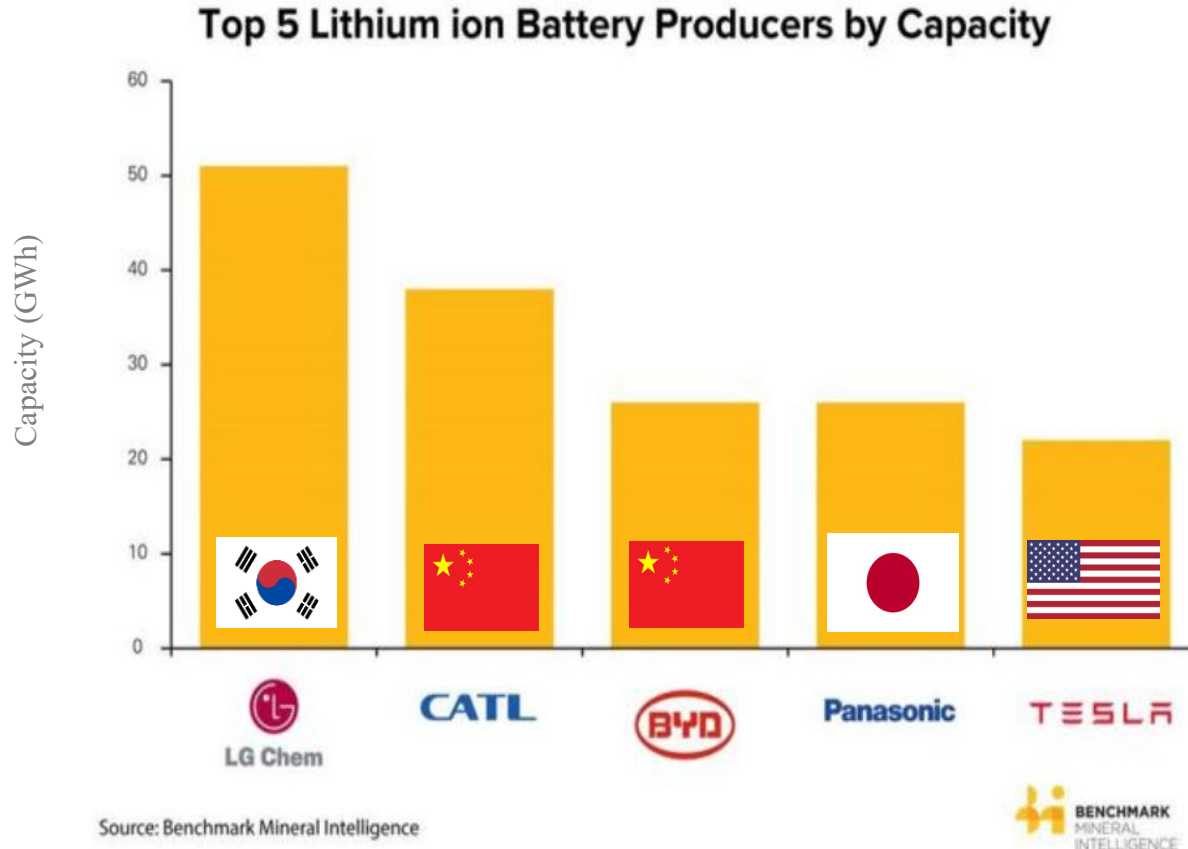


# Typical rechargeable battery configurations:

## A, coin, B, cylindrical, C, prismatic, and D, pouch shapes



# Production is increasing to meet demand



- Tesla Gigafactory in Nevada
  - 20 GWh production
  - Joint project with Panasonic
- Also Gigafactories in NY and Shanghai
- More factories planned

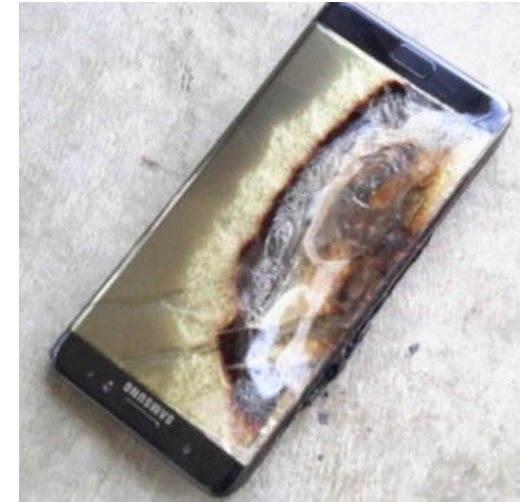
# Safety Concerns of Lithium-ion Batteries



Boeing 787, Dec. 2014 <sup>(1)</sup>



Tesla Model S, Aug. 2016 <sup>(2)</sup>



Samsung Note 7, Sept. 2016 <sup>(3)</sup>

- LIBs dominate rechargeable energy storage market due to high energy density
- Safety incidents still occurring for mature Li-ion battery technology
- Susceptible to thermal runaway: can occur by overcharging, cell puncture, dendrites

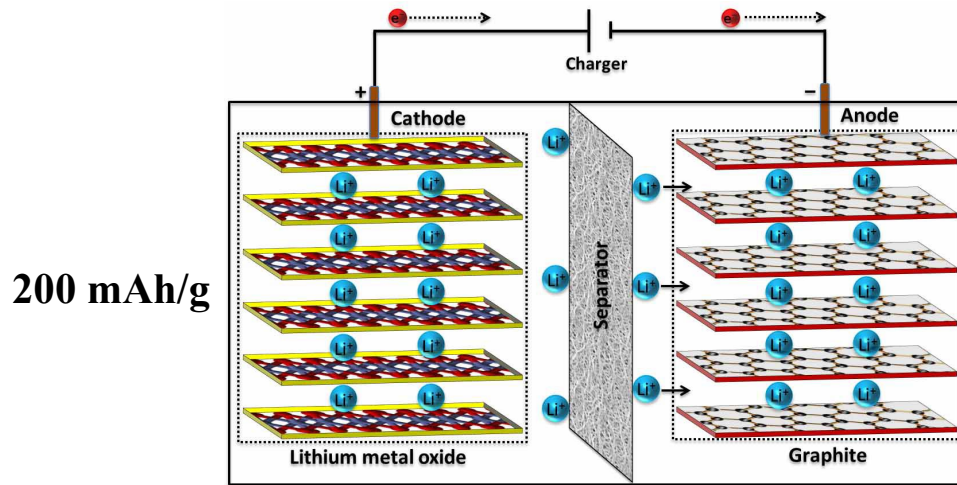
**Motivation: Improve understanding of thermal runaway and how to mitigate for rechargeable battery**

(1) <https://www.scientificamerican.com/article/how-lithium-ion-batteries-grounded-the-dreamliner/>

(2) <https://electrek.co/2016/08/15/tesla-model-s-catches-fire-test-drive-france/>

(3) <http://www.cbsnews.com/news/samsung-galaxy-note-7-batteries-fires-faa-warnings-passengers-worldwide-rec/>

**Need:** High safety, high energy density solid-state Li metal batteries required for electric vehicles, electronics and defense applications



➤ Graphite: 372 mAh/g



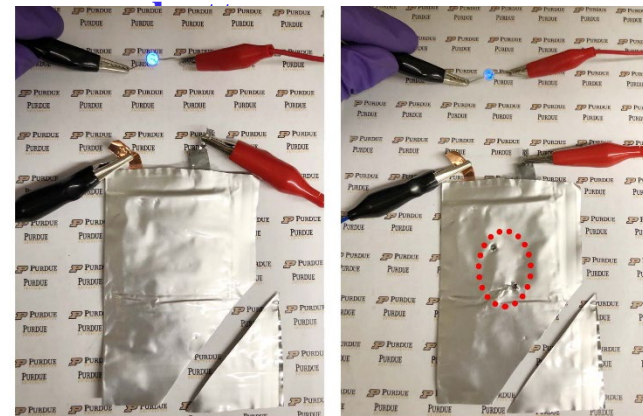
➤ Li metal: 3840 mAh/g

**Li-ion battery with liquid electrolytes**



Energy density ~ 250 Wh/kg

**Solid-state**



**Cutting**

**Punching**

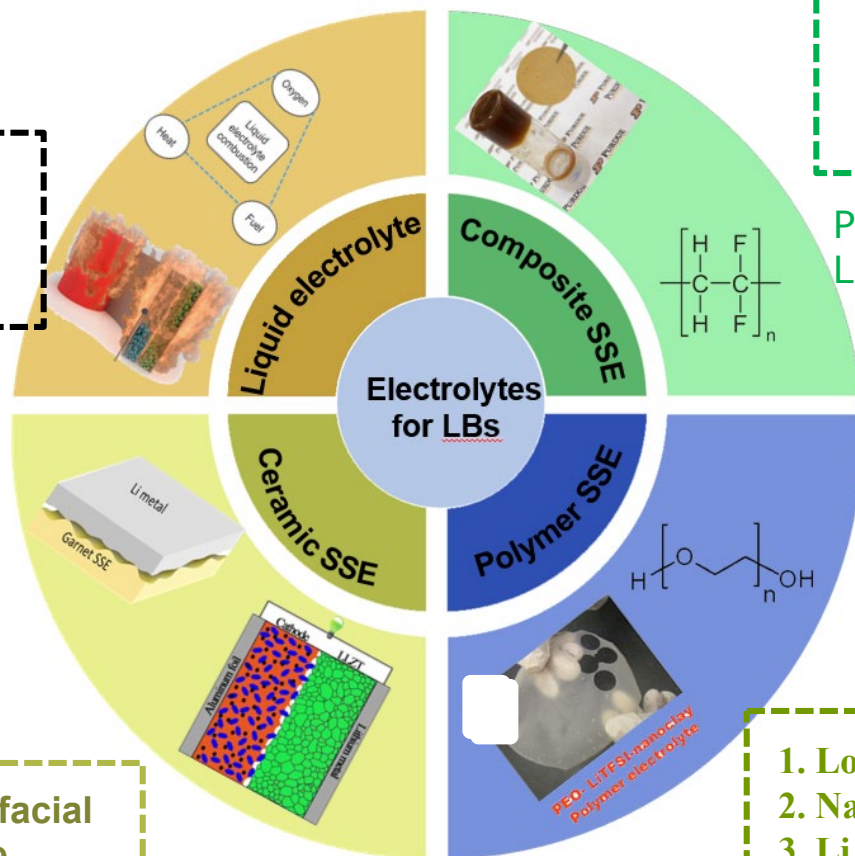
Energy density ~ 450 Wh/kg

# Purdue's Advanced Technology

## Our work



1. Flammable
2. Leak
3. Li dendrite



1. High ionic conductivity
2. Wide voltage window
3. Thermal Safe
4. Li Dendrite free



Polyvinylidene fluoride (**PVDF**)  
 $\text{Li}_{6.4}\text{La}_3\text{Zr}_{1.4}\text{Ta}_{0.6}\text{O}_{12}$  (**LLZTO**)



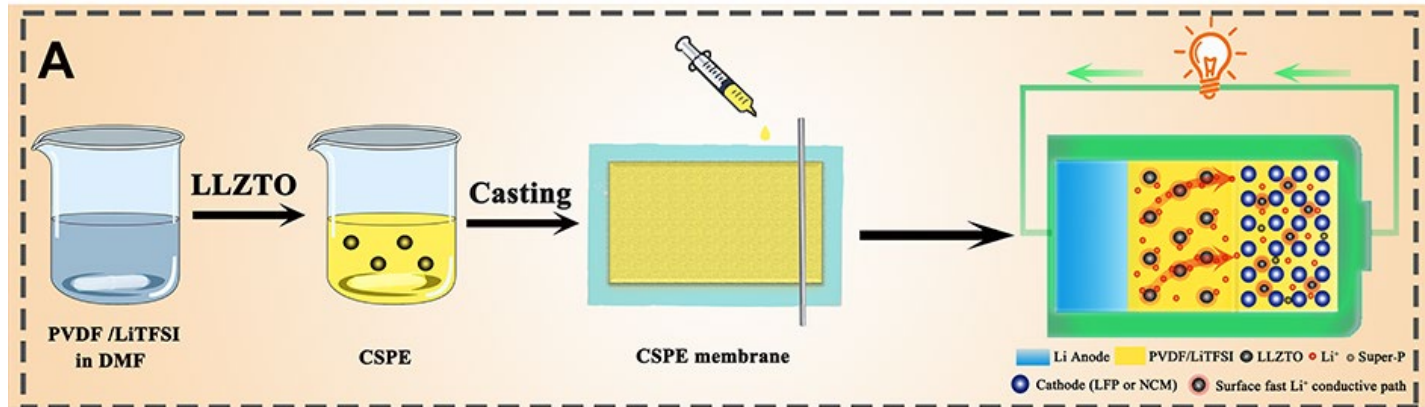
1. High interfacial resistance
2. Li dendrite

1. Low ionic conductivity
2. Narrow voltage window
3. Li dendrite



Poly(ethylene oxide) (PEO)

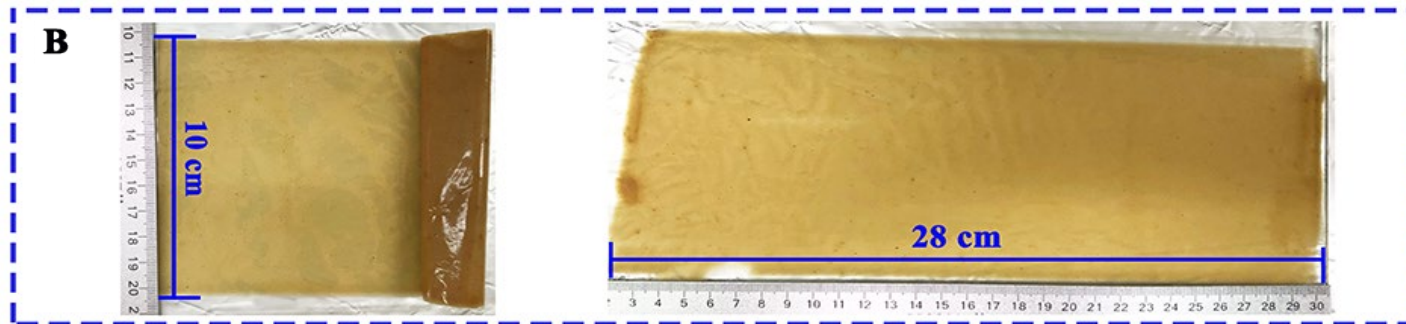
# Scalable Fabrication of SS Composite Electrolyte



✓ Facile synthesis

✓ Operation at room temperature

The synthesis of composite solid polymer electrolyte



Pictures of as-prepared composite solid polymer electrolyte

✓ Flexible

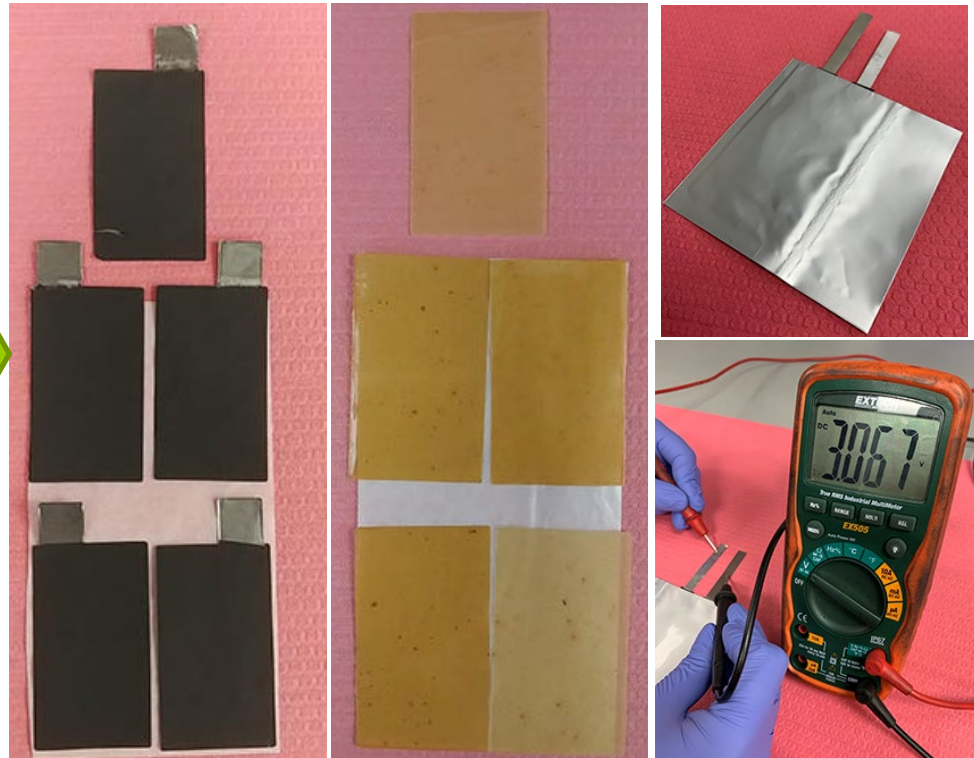
✓ Free standing

✓ Scalable

# A scalable solid state battery

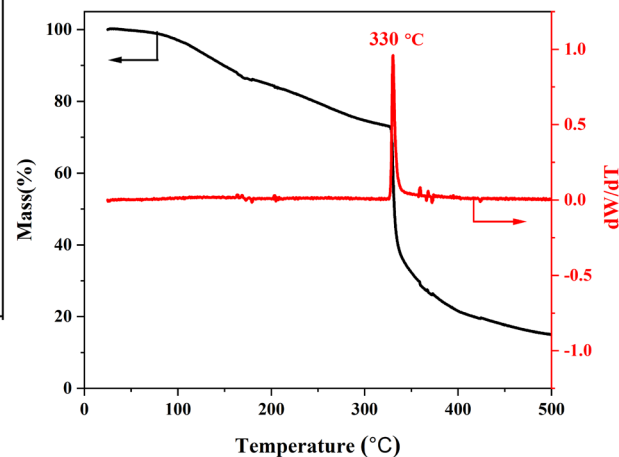
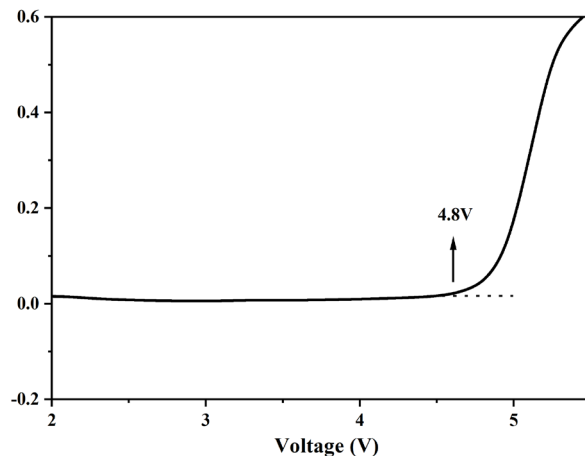
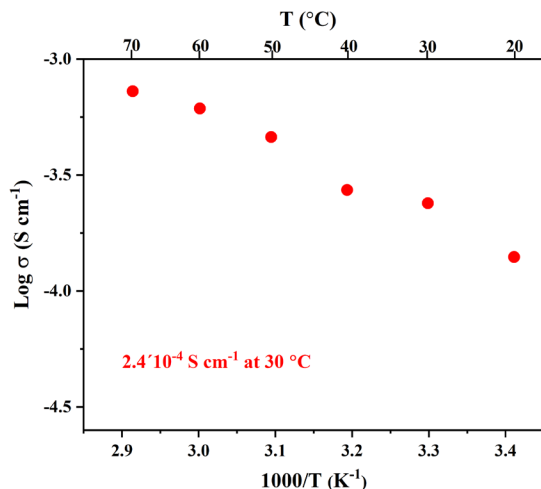


Small scale coin cell



Large scale, single-layer pouch cell

# Ionic Conductivity, Voltage Window, Thermal Stability



- ✓ High room-temperature ionic conductivity ( $2.4 \times 10^{-4} S \text{ cm}^{-1}$ )
- ✓ Wide voltage window ( $\sim 4.8 V$ )
- ✓ Excellent thermal stability ( $\sim 330^\circ C$ )



VS

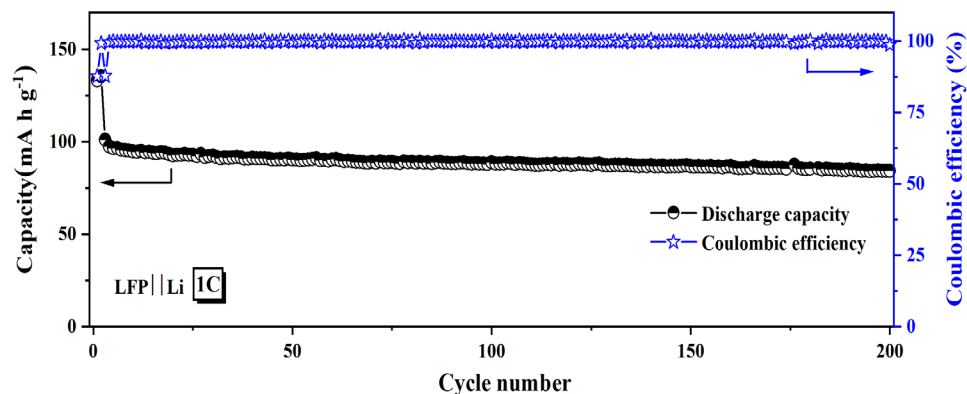


- ✗ Low room-temperature ionic conductivity ( $10^{-7} - 10^{-5} S \text{ cm}^{-1}$ )
- ✗ Narrow voltage window ( $\sim 3.8 V$ )
- ✗ Inferior thermal stability ( $\sim 230^\circ C$ )

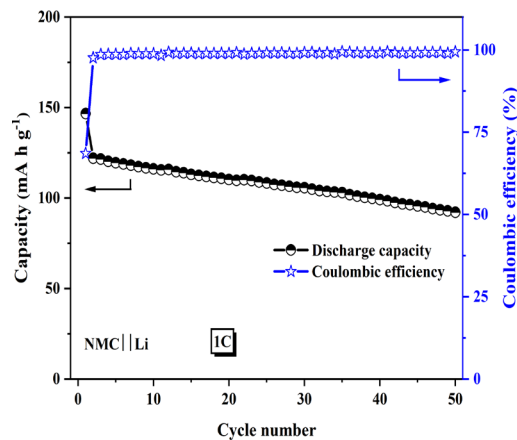
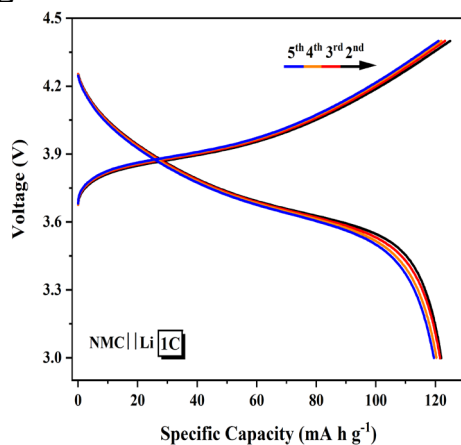
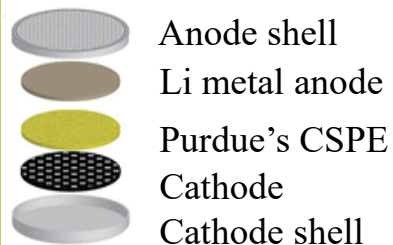
Purdue's Composite solid polymer electrolyte

Typical PEO-based polymer electrolyte

# Electrochemical Performance of Solid-state Full Cell



**LiFePO<sub>4</sub>**  
2.8 ~ 3.8 V



**Room Temperature**

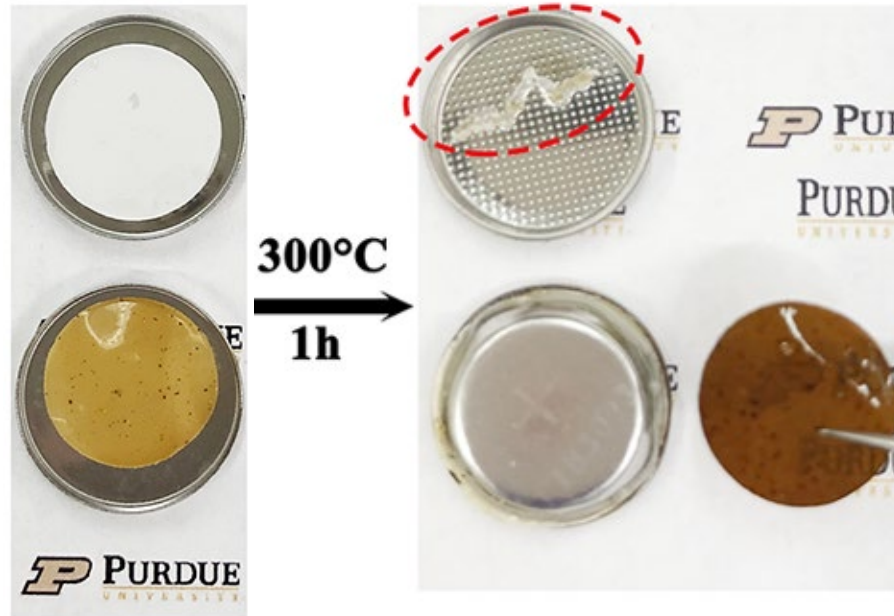
**LiNi<sub>1/3</sub>Mn<sub>1/3</sub>Co<sub>1/3</sub>O<sub>2</sub>**  
3 ~ 4.4 V



# Thermal Stability

**Commercial PP separator**

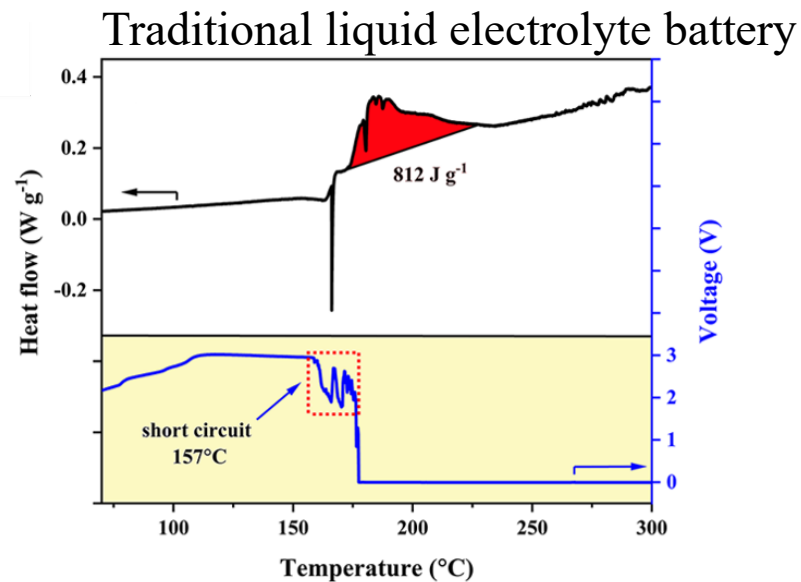
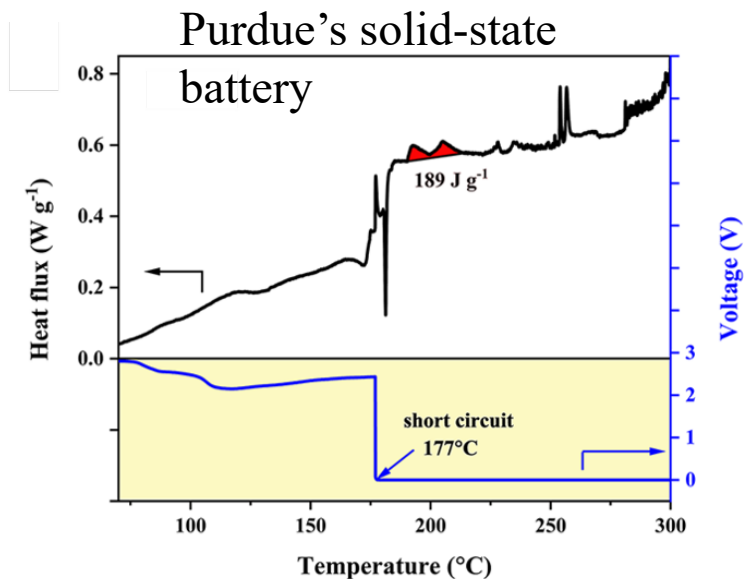
**Purdue's CSPE**



**× Melt and shrink**

**✓ Maintain its structure**

# Thermal Safety Performance



Thermal stable window:

Heat generation:

✓ up to 177  
 $^{\circ}\text{C}$



VS



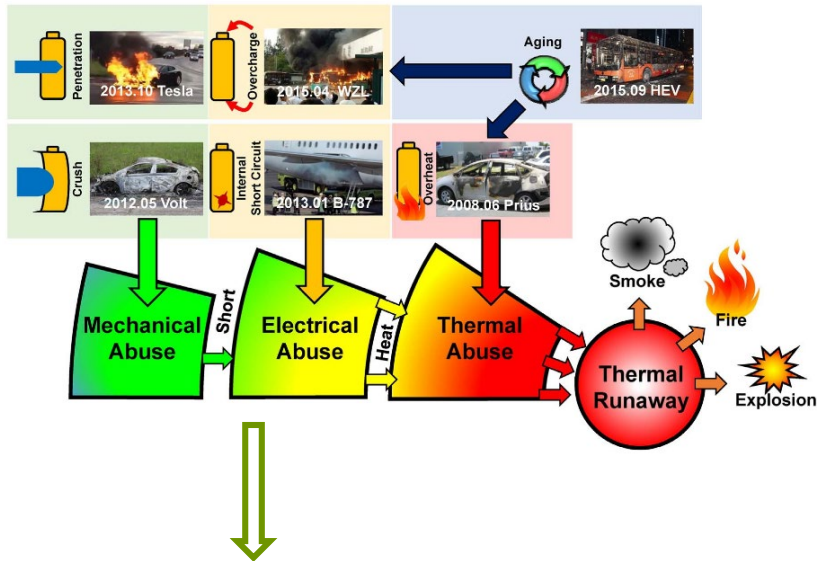
✗ up to 157  
 $^{\circ}\text{C}$

✓ 189  $\text{J g}^{-1}$

✗ 812  $\text{J g}^{-1}$

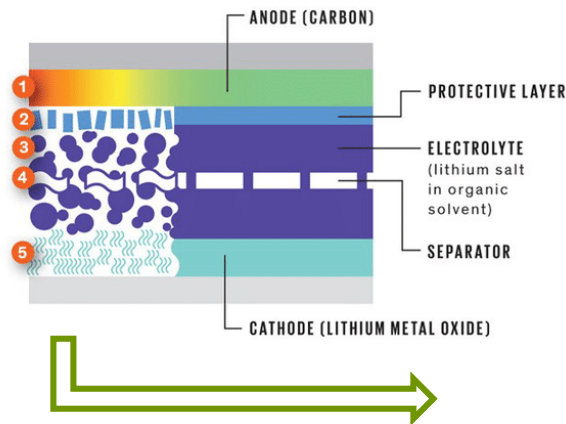
# Conventional Li-ion- Can we Sense the VOCs and Gases Early?

## Causes of Thermal Runaway

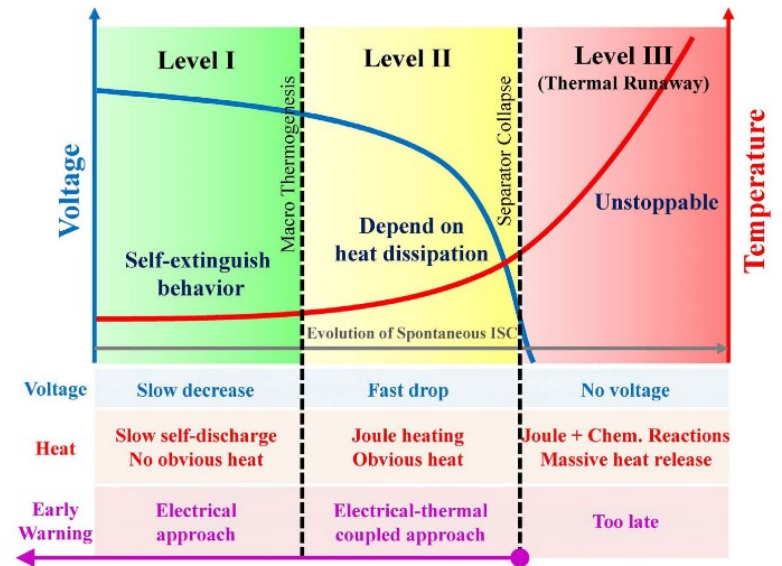


## Stages of Thermal Runaway

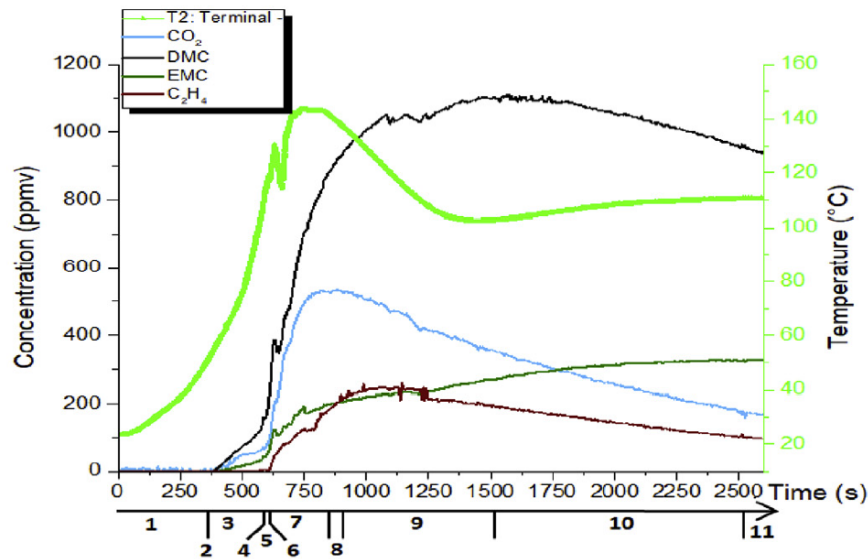
1. Heating starts.
2. Protective layer breaks down.
3. Electrolyte breaks down into flammable gases.
4. Separator melts, possibly causing a short circuit.
5. Cathode breaks down, generating oxygen.



## Level of Internal Short Circuit (ISC)

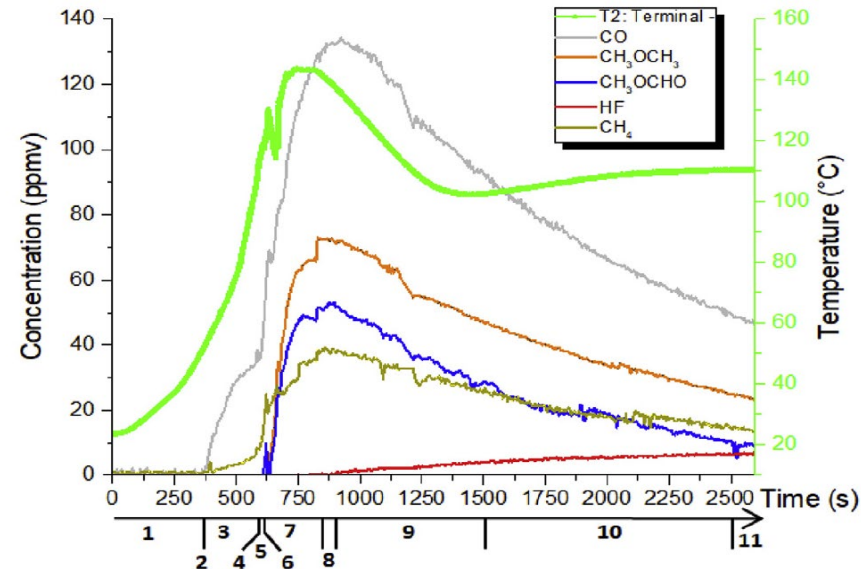


# Gases Released from Li-ion Batteries During Thermal Runaway



Gases Released from Batteries:

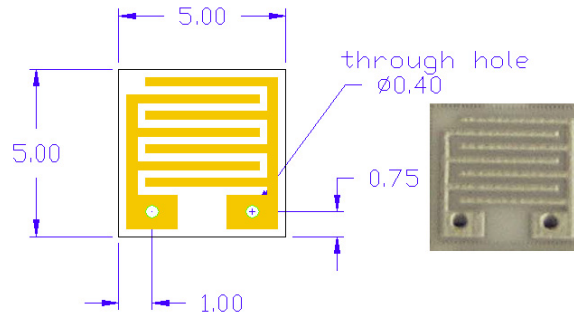
- Carbon Dioxide
- Carbon Monoxide
- Hydro-fluoride



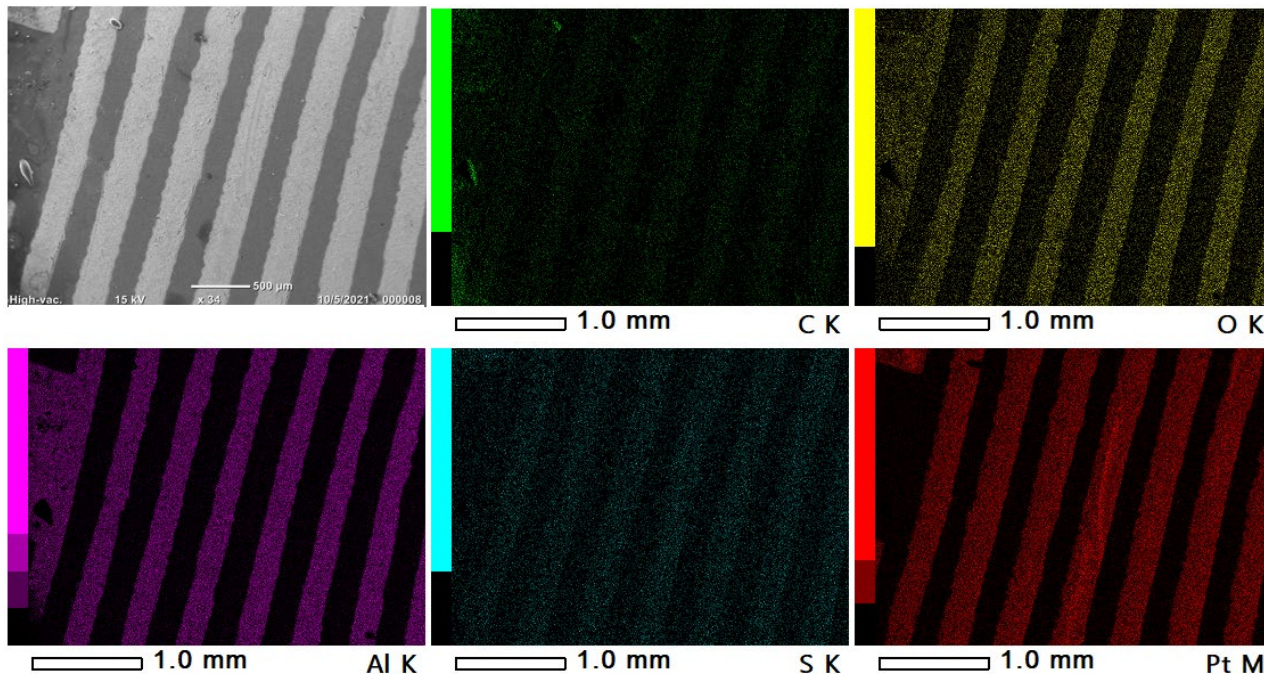
VOCs Released from Batteries:

- Di-methyl Carbonate
- Ethyl methyl carbonate
- Ethylene
- Dimethyl ether
- Methyl formate

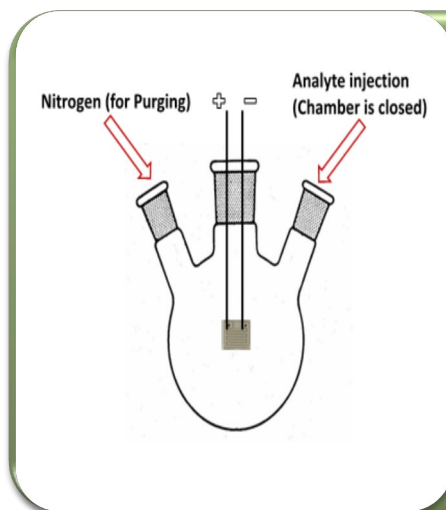
# PEDOT:PSS (poly(3,4-ethylenedioxythiophene polystyrene sulfonate) Sensor



- Interdigitated-Platinum Electrode
- Spin Coated
- Energy dispersive X-ray spectroscopy (EDS)
- Uniform coating of C, S and O indicate presence of PEDOT:PSS



# Impedance Response of Sensor



## Parameters & Observations:

Instrument: Gamry 600+

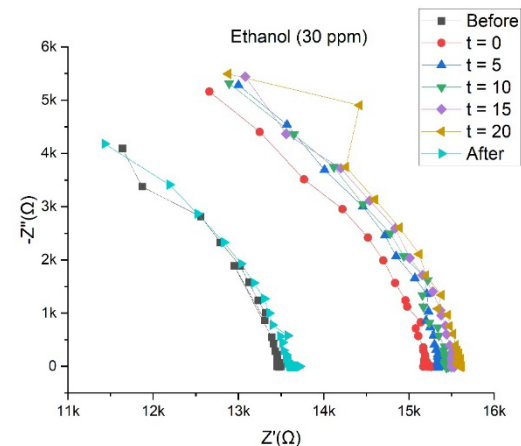
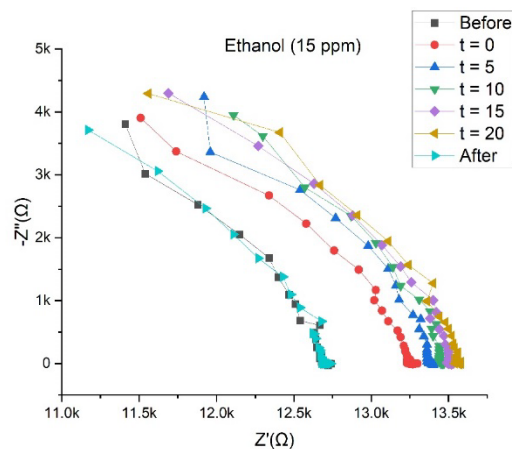
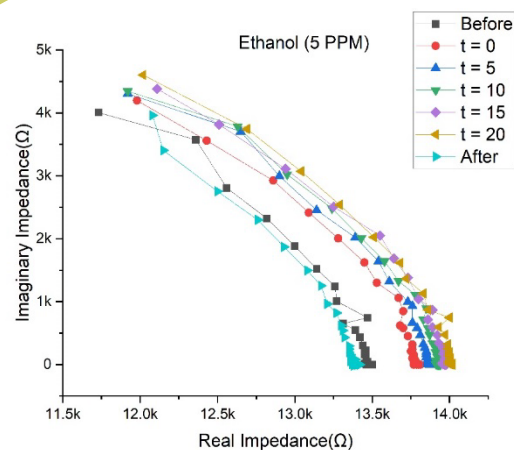
Frequency: 1 Hz- 1MHz

Samples: 5 samples/Analytes

Room Temperature

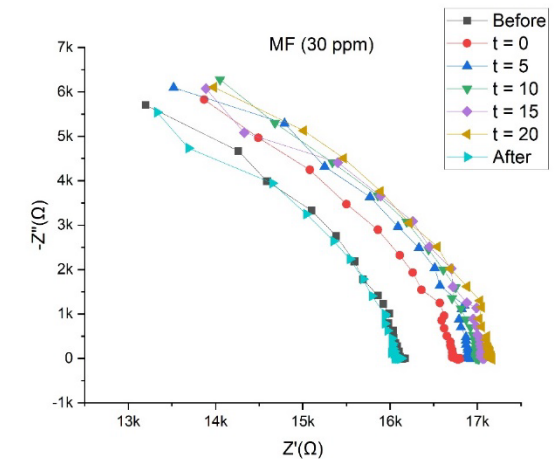
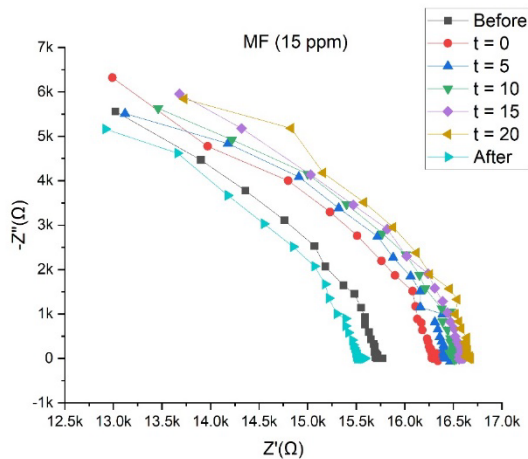
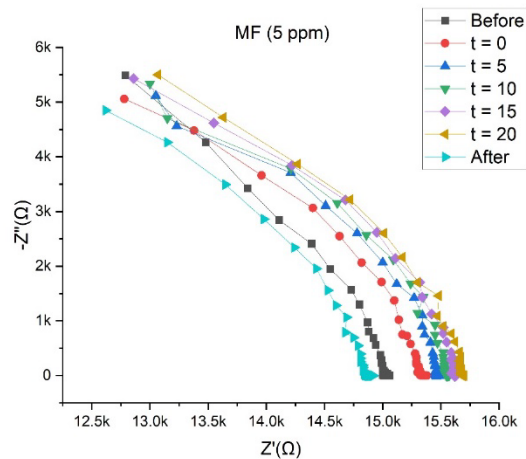
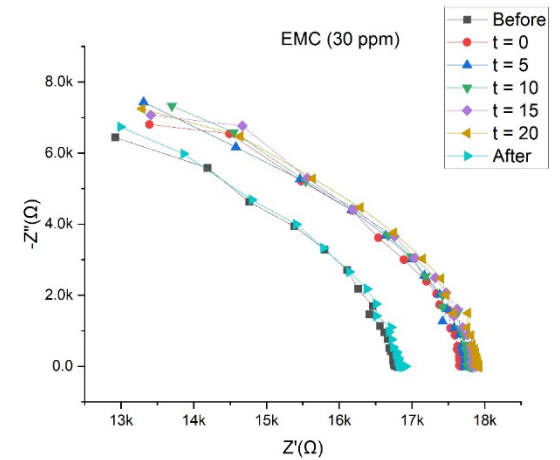
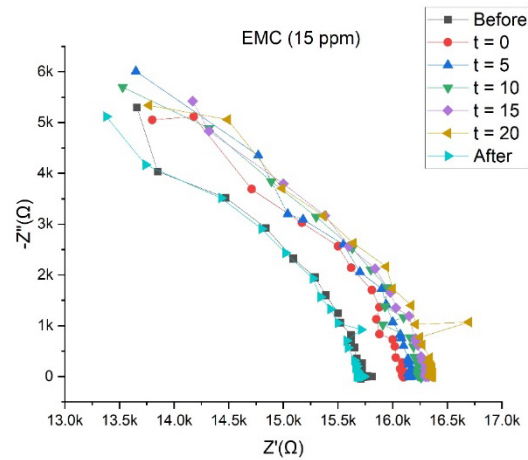
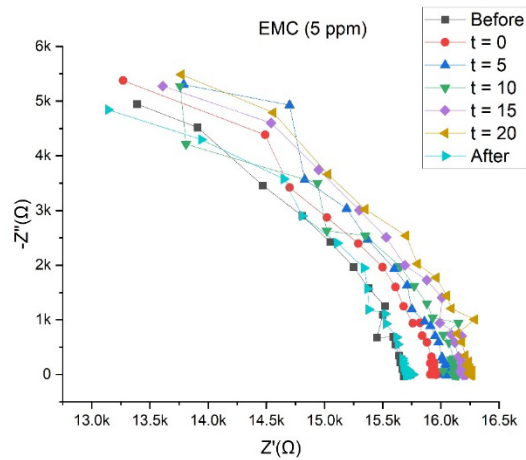
Mechanism: Adsorption/Desorption

Fast Recovery

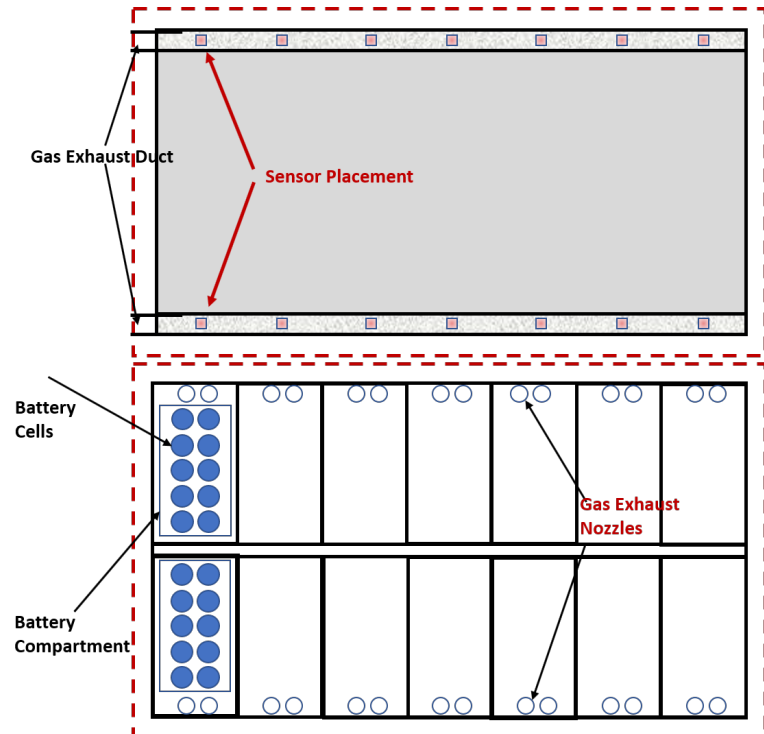
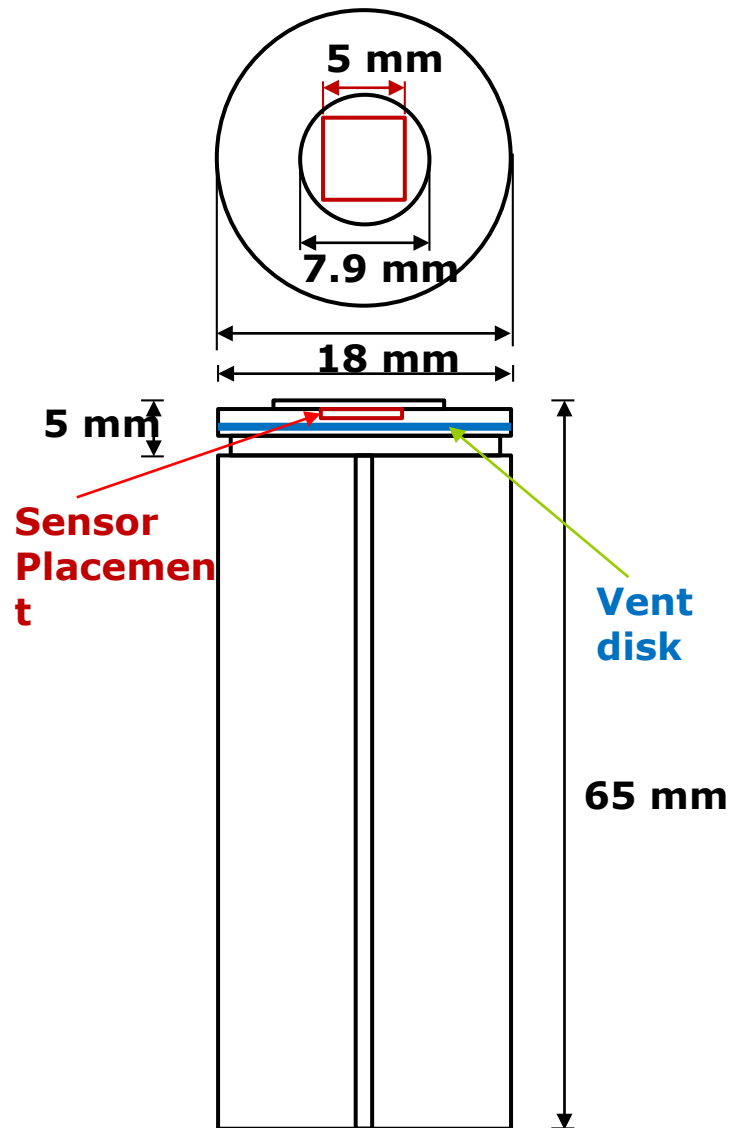


# Impedance Response to VOCs

(EMC and MF released as the byproduct of decomposition of Electrolytes)



# Applications of developed Chemosensor



- ✓ EV charging station,
- ✓ Private parking area,
- ✓ Battery storage rooms,
- ✓ Military applications

# Summary

1. Engineered composite solid state batteries could be **safer** than conventional Lithium ion batteries.
2. Li-metal batteries are **safe till 150 °C**, separator and lithium metal melts after that causing huge exothermic heat.
3. **Early chemosensing** could stop the catastrophic runaway of Li-ion batteries
4. **Chemosensors** can be applied to various places including parking garages, battery storage area, charging stations etc.

*Thank You!*

